

Recuerda:

Ley de los gases ideales:  $PV$

Constante de los gases:  $0.082 \text{ L atm mol}^{-1} \text{ K}^{-1}$

Constante de Avogadro =  $6.023 \times 10^{23} \text{ mol}^{-1}$

1 atm = 760 mm Hg

1 m<sup>3</sup> = 1000 L

°C = K - 273.15



Tabla Periódica de los Elementos de la IUPAC

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